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IS : 11774 - 1986

Indian Standard
METHOD FOR
DETERMINATION OF DICHLORVOS
RESIDUES IN FOOD COMMODITIES

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MANAK BHAVAN, 9 BAHADUR SHAH ZAFAR MARG
NEW DELHI 110002

Indian Standard

METHOD FOR

DETERMINATION OF DICHLORVOS RESIDUES IN FOOD COMMODITIES

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Indian Standard

METHOD FOR DETERMINATION OF DICHLORVOS RESIDUES IN FOOD COMMODITIES

0. FOREWORD

0.1 This Indian Standard was adopted by the Indian Standards Institution on 18 September 1986, after the draft finalized by the Pesticides Residue Analysis Sectional Committee had been approved by the Agricultural and Food Products Division Council.

0.2 Dichlorvos is used in agriculture for the control of insect pests. Assessment of residues is, therefore, an important step in safeguarding human health.

0.3 In the preparation of this standard, due consideration has been given to the limits of dichlorvos laid under the provisions of **Prevention of Food Adulteration Rules**, 1965 and the test methods are sensitive to the prescribed levels of residues.

0.4 This standard will enable the health authorities **and others engaged** in the field to follow a uniform test procedure for the estimation of dichlorvos residues in various food commodities.

0.5 In reporting the result of a test or analysis made in accordance with this standard, if the final value, observed or calculated, is to be rounded off, it shall be done in accordance with IS : 2-1960*.

1. SCOPE

1.1 This standard prescribes the gas chromatographic method (GLC) for the determination of dichlorvos residues in food commodities.

2. QUALITY OF REAGENTS

'2.1 Unless specified otherwise, pure chemicals and distilled water (see IS : 1070-1977†) shall be employed in tests.

NOTE—'Pure chemicals shall mean chemicals that do not contain impurities which affect the results of analysis.

*Rules for rounding off numerical values (*revised*).

†Specification for water for general laboratory use (*second revision*).

3. SAMPLING

3.1 The representative samples for the purposes of estimating dichlorvos residues in foods shall be drawn in accordance with IS : 11380 (Part 1)-1985*.

4. PREPARATION OF THE LABORATORY SAMPLE, CLEAN UP AND EXTRACTION

4.1 Apparatus

4.1.1 Waring Blender or Equivalent

4.1.2 Kuderna Danish Evaporative Concentrator

4.1.3 Water-Bath

4.1.4 Chromatographic Column — 25 cm long and 2 cm i.d.

4.2 Reagents

4.2.1 Methylene Chloride — AR grade, glass redistilled.

4.2.2 n-Hexane — AR grade, glass redistilled.

4.2.3 Ethyl Acetate — adsorption grade, glass redistilled.

4.2.4 Sodium Sulphate — anhydrous.

4.3 Extraction

4.3.1 Fruits and Vegetables — Shred the entire laboratory sample. Macerate a subsample of 25 g with 100 ml of methylene chloride using the **waring** blender. Filter the suspension under slight vacuum through a **6-cm** buchner funnel layered with suitable filter paper. Rinse, jar and filter cake twice with 50 ml portions of methylene chloride. Transfer the methylene chloride extract to a separatory funnel, and drain off the organic layer into **500-ml** erlenmeyer flask. Dry with 4-5 g of anhydrous sodium sulphate, and filter through a folded paper into a **500-ml** round-bottom flask. Concentrate the methylene chloride solution to about 20 ml and proceed as described under 4.4.

4.3.2 Extraction of Oily Crops — Place 25 g of milled grain into a soxhelt thimble and extract with 300 ml of methylene-chloride in a soxhlet apparatus for **4** hours, using a **500-ml** round bottom flask. Concentrate the extract to approximately 20 ml by using the rotating evaporator (**35-40°C**) and add 50 ml of hexane. Concentrate again, and add a further portion of 50 ml hexane. Concentrate second time to a volume of **10-15** ml and transfer the hexane extract to a **50-ml** volumetric flask. Wash the evaporation flask twice with small portions of hexane, add the **washings** to the extract and make up to volume with hexane.

*Methods of sampling for determination of pesticides residues : Part 1 Agricultural and food commodities.

4.4 Clean-Up — Extract the hexane solution (50 ml) four times with 100 ml of distilled water in a 250-ml separatory funnel. Combine the aqueous extracts and re-extract dichlorvos from the water with two 100-ml portions of ethyl acetate using 1-litre separatory funnel. Collect the organic solutions in an erlenmeyer flask, dry with 4-5 g anhydrous sodium sulphate, and filter through a folded paper into a round-bottom flask. Wash the erlenmeyer flask and the sodium sulphate with 50 ml of hexane, which portion is added to the ethyl acetate extract. Concentrate on the rotating evaporator (35-40°C) to a volume of about 5 ml, and transfer the concentrate to a 10-ml glass-stoppered graduated test tube. Rinse evaporation flask with 3-4 ml of hexane, add the washing to the concentrate, and make up to 10 ml.

5. GAS CHROMATOGRAPHIC METHOD

5.1 The extracted dichlorvos is measured gas chromatographically using a NPD detector or a flame photometric detector (phosphorus specific). The content of dichlorvos is determined by comparing the response with reference standard of dichlorvos.

5.1.1 Gas Chromatograph — A gas chromatograph equipped with a NPD detector or a flame photometric detector operated under the following suggested parameters. These parameters may be varied according to the available facilities provided standardization is done:

Column	Glass, 2 ml long and 2 mm i.d. packed with 10 percent OV 210 on Gas Chrom Q (80-100 mesh)
Column oven temperature	130°C
Injection port temperature	200°C
Detector temperature	200°C
Carrier gas and flow rate	Nitrogen 40 Hydrogen 15-20 } Air 150 J ml per minute
Retention time	5 mts
Chart speed	5 mm/mt
Attenuation	100 x 1

5.2 Procedure — Inject suitable aliquot (1 μ l) into the column using a microlitre syringe. Identify the peak by its retention time and measure the peak area. For reference standard prepare hexane solutions containing one to ten ppm of dichlorvos.

5.3 Calculation

$$\text{Dichlorvos residue, } \mu\text{g/g (ppm)} = \frac{A_1 \times V_2 \times V_3 \times C \times f}{A_2 \times V_1 \times M}$$

where

A_1 = peak height/area of the sample;

V_2 = volume, in μl , of standard dichlorvos injected;

V_3 = total volume, in ml, of sample solution;

C = concentration, in $\mu\text{g/g}$, of standard dichlorvos solution;

f = recovery factor = $\frac{100}{\text{percent mean recovery}}$.

A_2 = peak height/area of the dichlorvos standard;

V_1 = volume of sample in μl injected; and

M = mass, in g, of sample taken for analysis.